

REMARKS

Claims 1-23, 26-28 and 30 are pending in the application and stand rejected. Applicants respectfully request reconsideration in view of the following remarks.

Rejections under 35 U.S.C. § 112

Claims 1-23, 26-28 and 30 are rejected under 35 U.S.C. § 112, second paragraph, as allegedly being indefinite. The Office has alleged that steps leading to the formation of different products (i.e., micelles, vesicles, emulsion, gel and matrix) are missing in the claim. In particular, the Office alleges that the specification does not show how vesicles are formed. *See* Office Action at page 2. Applicants respectfully traverse the rejection.

Claim language is definite when the metes and bounds of the invention can be adequately determined. *In re Goffe*, 526 F.2d 1393, 1397 (CCPA 1975); *see* MPEP § 2173.02. The Office has alleged that the claims are indefinite because the specification does not show how vesicles are formed. Applicants respectfully disagree.

The requirement under 35 U.S.C. § 112, second paragraph is evaluated in the context of whether the scope of the claim is clear to one possessing the ordinary level of skill in the art. MPEP § 2171. The formation of vesicles was well known in the art at the time of the invention. For a review, *see* Kozubek, et al. (Acta Biochimica Polonica (2000); 47:639-649; Exhibit A) at page 640-641, describing the characteristics and preparation of liposomal vesicles). Thus, one of ordinary skill in the art would readily understand the phrase “vesicles” as claimed. In view of this, the metes and bounds of the rejected claims are sufficiently clear. Accordingly, the present rejection should be withdrawn.

The Office has also alleged that the specification does not provide a distinction between the terms “gels” and “extended networks”. Applicants note that only the term “gel” appears in the claims. The specification describes gel forming copolymers generally, and poloxamer gels specifically, at, for example, page 34, lines 24-28 and page 35, line 30 through page 36, line 4,

respectively. Further, the specification describes that, at higher concentrations, poloxamers have a tendency to undergo gel formation under certain temperature conditions. *See* specification at page 33, lines 10-12. Accordingly, one of skill in the art would understand the scope of the term “gel” as claimed.

Because the claims as amended are clear and definite, Applicants respectfully request that the rejection under 35 U.S.C. § 112, second paragraph be withdrawn.

Rejections under 35 U.S.C. § 103

The Office has maintained the rejection of claims 1-23, 26-28 and 30 under 35 U.S.C. § 103(a), as allegedly being unpatentable over Schneider (6,258,378), alone or in combination with Lyons (5,616,342) and Young (6,375,930). Claims 1-23, 26-28 and 30 are also rejected under 35 U.S.C. § 103(a), as allegedly being unpatentable over Lyons, in combination with Klaveness (5,674,468) and/or See (6,015,576). Furthermore, claims 1-23, 26-28 and 30 are rejected under 35 U.S.C. § 103(a), as allegedly being unpatentable over Lyons, in view of Desai (6,074,666) or Madden (5,389,378) in further combination with Unger (6,028,066). Applicants traverse the rejections.

To establish a *prima facie* case of obviousness, a three-prong test must be met. First, there must be some suggestion or motivation, either in the references or in the knowledge generally available among those of ordinary skill in the art, to modify the reference. *In re Rouffet*, 149 F.3d 1350, 1357 (Fed. Cir. 1998). Second, there must be a reasonable expectation of success found in the prior art. *In re Vaeck*, 947 F.2d 488 (Fed. Cir. 1991). Third, the prior art references must teach or suggest all the claim limitations. *In re Royka*, 490 F.2d 981, 985 (CCPA 1974). MPEP § 2143.

Claims 1-23, 26-28 and 30 are rejected under 35 U.S.C. § 103(a), as allegedly being unpatentable over Schneider (6,258,378), alone or in combination with Lyons (5,616,342) and Young (6,375,930). The Office acknowledges that Schneider does not disclose formulations containing a photosensitizer. However, the Office asserts that it would have been obvious to incorporate a photosensitizer as the therapeutic agent in the liposomal formulations disclosed by

Schneider, with a reasonable expectation of success. *See* Office Action at pages 3-4. In particular, the Office has alleged that Schneider exemplifies the use of Pluronic® F-108 and states that Pluronic® and Poloxamer® surfactants could be used, and therefore it would have been obvious to one of ordinary skill in the art to select the appropriate poloxamer. *See* Office Action at page 7, citing Schneider at col. 7, lines 1-3 and col. 13, line 11. Applicants respectfully disagree.

Read in context, the portions of Schneider cited by the Office do not support the assertion that the reference suggests the use of poloxamers as in the claimed invention. Rather, Schneider states that “[a]s the surfactants or detergents were only used to facilitate the lipid solubilization and gas microbubble formation, they were removed after the microbubble formation.” *See* Schneider at col. 13, lines 6-8 (emphasis added); *see also* Schneider at col. 12, lines 60-64. Schneider neither teaches nor suggests the formation of a dried composition comprising a poloxamer carrier and a therapeutic agent physically associated with a solid support, let alone a dried polypyrrolic photosensitizer-carrier composition as claimed.

The Office alleges that one of skill in the art would have been motivated to use Schneider’s formulations to deliver photosensitizers in view of the disclosures of Lyons and Young, which the Office characterizes as showing the routine practice of using poloxamer containing emulsions, micelles and liposomes for the delivery of photosensitizers. *See* Office Action, page 3-4.

The Office asserts that Lyons discloses emulsion formulations containing photosensitizers and block copolymers such as poloxamers. *See* Office Action at page 4, citing Lyons at abstract, col. 3, line 10 through col. 8 line 22, and claims. Applicants respectfully note that the abstract is silent as to the nature of the surfactant, and the claims refer to formulations containing a phospholipid stabilizer and a required bile acid costabilizer. The only mention of poloxamers describes their potential use as emulsion stabilizers, and notes that the preferred stabilizer is egg yolk phospholipid. *See* Lyons at col. 8, lines 17-22.

As previously discussed in responses of record, Lyons discloses oil-in-water emulsions comprising a dispersion of oil droplets in a continuous aqueous phase with a surfactant used to

stabilize the dispersed droplets. See Lyons at col. 2, lines 30-34. As noted above, the only reference to poloxamers is as potential emulsion stabilizers. As with Schneider, the surfactants in Lyons are thus distinguishable from the claimed carriers, which are combined with a polypyrrolic photosensitizer to form a dried photosensitizer-carrier composition physically associated with a solid support, wherein the complex may form an emulsion upon hydration.

The Office also cites Lyons at the abstract, col. 4, lines 44-65 and Example 1 to support the assertion that Lyons discloses emulsion formulations containing photosensitizers and poloxamers, and specifically Pluronic® F-127. See Office Action at pages 4 and 7. Applicants are unable to locate the alleged reference to Pluronic® F-127. Specifically, the abstract is silent as to the nature of the surfactant; col. 4, lines 44-65 are blank; and Example 1 describes the use of egg phospholipid as the surfactant. Applicants respectfully request that the Examiner provide the column and line citation for the use of Pluronic® F-127 so that applicants can adequately respond.

The Young reference is silent regarding the use of copolymers at all. Thus, even if combined, the combination fails to teach all the elements of the claimed invention. In particular, the combination fails to teach a dried photosensitizer-carrier complex, comprising a mixture of a polypyrrolic macrocyclic photosensitizer and a poloxamer copolymer carrier physically associated with a solid support.

To render the claimed invention obvious, there must be a motivation to combine the cited references. If a proposed modification would render the prior art invention being modified unsatisfactory for its intended purpose, then there is no suggestion or motivation to make the proposed modification. *In re Gordon*, 733 F.2d 900, 221 USPQ 1125 (Fed. Cir. 1984); MPEP § 2143.01(V).

Lyons discloses that the presence of bile acid salts as costabilizers is crucial to achieving the desired results. See Lyons at col. 8, lines 27-29. The claimed formulations lack a bile acid salt. Because the deletion of the bile acid salts would render the Lyons invention unsatisfactory for its

intended purpose, there is no motivation to combine Lyons with Schneider and Young. MPEP § 2143.01(V).

Schneider describes the removal of the surfactant or detergent (see col. 13, lines 6-8), and further describes that “size distribution and microbubble number may be equally tailored by controlling the duration of decantation ...” (i.e. the process used to remove the surfactant). See Schneider at col. 12, lines 60-67. Schneider provides no motivation that would lead one of skill in the art to form a dry complex wherein the surfactant has not been removed.

Failure to remove the surfactant as described would change the principle of operation of Schneider’s invention. It is well-settled law that if a proposed modification or combination would change the principle of operation of the prior art invention being modified, the teachings of the references are not sufficient to render the claims *prima facie* obvious. *In re Ratti*, 270 F.2d 810, 123 USPQ 349 (CCPA 1959); MPEP § 2143.01(VI).

Moreover, even if combined, there is no reasonable expectation of success based on the combination of Schneider, Lyons and Young. One of skill in the art would not expect the claimed formulations, which lack the crucial bile acid salts required by Lyons, and contain surfactants which are removed by Schneider, to produce emulsions useful for the delivery of photosensitizers. Thus, the combination of Schneider with Lyons and Young may only be based on improper hindsight reasoning, taking account of knowledge not within the ordinary skill in the art at the time of the invention.

Further, block copolymers were not expected to be useful as such because of the greater difficulty in controlling and maintaining particle size during manufacture and storage. See specification at page 47, lines 24-28.

Based on the above, claims 1-23, 26-28 and 30 are nonobvious under Schneider, alone or in combination with Lyons and Young. Applicants respectfully request that the rejection be withdrawn.

Claims 1-23, 26-28 and 30 are also rejected under 35 U.S.C. § 103(a), as allegedly being unpatentable over Lyons, in combination with Klaveness (5,674,468) and/or See (6,015,576). The Office acknowledges that Lyons does not disclose the preparation of the photosensitizer-copolymer carrier composition in a dried form in the presence of a solid support such as lactose. These deficiencies are not remedied by combination with Klaveness and/or See.

As discussed above, Lyons describes the use of poloxamers only as potential emulsion stabilizers and requires the presence of bile acid salts as costabilizers. The Office asserts that it would have been obvious to prepare the emulsions of Lyons in dried form using lactose as a solid support, in view of Klaveness and See. The Office cites Klaveness (at col. 40, lines 28-45) and See (at abstract; col. 6, line 57 through col. 7, line 8) as disclosing lyophilization of emulsions in the presence of cryopreservatives, such as lactose. Applicants respectfully disagree.

The portion of Klaveness cited by the Office does not describe lyophilization of an emulsion containing a poloxamer in the presence of a cryopreservative. *See* Klaveness at col. 40, lines 28-45. Klaveness describes the isolation of polymeric beads by filtration from a naphthalene-water emulsion containing Pluronic® F68. *See* Klaveness at col. 40, lines 36-39. The beads were resuspended in water which contained lactose, and the aqueous suspension was lyophilized. *See* Klaveness at col. 40, lines 41-45; *see also* Klaveness at col. 13, lines 21-29 (describing generally the lyophilization of particulate polymeric material precipitated and collected from emulsions containing polymeric materials, such as poloxamers). See describes lyophilization of an emulsion containing liposomal antigens, but does not disclose emulsions containing poloxamers at all. *See* See at col. 6, line 57 though col. 7, line 8. Thus the combination of Lyons with Klaveness and/or See does not teach all the elements of the claimed invention.

Further, there is no motivation to combine the cited references. A prior art reference must be considered in its entirety, i.e., as a whole, including portions that would lead away from the claimed invention. *W.L. Gore & Associates, Inc. v. Garlock, Inc.*, 721 F.2d 1540, 220 USPQ 303 (Fed. Cir. 1983), *cert. denied*, 469 U.S. 851 (1984); MPEP § 2141.02(VI). As already discussed, Lyons discloses that the presence of bile acid salts as costabilizers is crucial to achieving the desired

results. *See* Lyons at col. 8, lines 27-29. Thus, Lyons teaches away from the claimed invention, which lacks the required bile acid salts. Based on the disclosure of Lyons, one of skill in the art would have no motivation to combine the cited references that would lead to the invention as claimed.

See describes emulsions containing viral antigens and lacks the poloxamer required by the instant claims. Such viral antigens are clearly distinguishable from the polypyrrolic photosensitizers of the present invention. Moreover, the present invention lacks the bile acid salts required by Lyons. Thus, one of skill in the art would not have had a reasonable expectation of success based on the combination of Lyons and See.

The Office asserts that one of skill in the art would expect similar results on lyophilization of an emulsion irrespective of the components. Applicants respectfully disagree, and request that the Examiner provide factual support for this assertion in the form of a reference. Alternatively, if the Examiner is relying on facts within his personal knowledge, Applicants respectfully request that the Examiner provide an affidavit to support these facts, so that the record may be made complete for appeal. MPEP § 2144.03.

Based on the discussion above, as well as the reasons already of record, claims 1-10, 16-28 and 30 are nonobvious under Lyons in combination with Klaveness and/or See. Applicants respectfully request that this rejection be withdrawn.

Claims 1-23, 26-28 and 30 also stand rejected under 35 U.S.C. § 103(a), as allegedly being unpatentable over Lyons, in view of Desai (6,074,666) or Madden (5,389,378) in further combination with Unger (6,028,066).

The Lyons patent has been discussed in detail above. The failure of Lyons to teach the invention as claimed is not remedied by the combination of Desai or Madden, in further combination with Unger. The Desai and Madden references are entirely silent regarding poloxamers. Unger only describes the use of poloxamers as emulsifying and/or solubilizing agents to stabilize gaseous precursor filled vesicles. *See* Unger at col. 40, lines 41-55 through col. 41 line

12. Thus, the combination of references does not teach all the elements of the invention as claimed. Specifically, the combination fails to teach a dried photosensitizer-poloxamer carrier composition physically associated with a solid support.

As already described, Lyons teaches away from the claimed invention, which lacks the bile acid salt costabilizer that is described as crucial to Lyons' invention. *See* Lyons at col. 8, lines 27-29. Thus, Lyons provides no motivation to combine the cited references that would lead to the present invention.

Given the lack of the bile acid salts required by Lyons in the claimed invention one of skill in the art would not have had a reasonable expectation of success based on the combination of Lyons with Desai or Madden, in further combination with Unger. Furthermore, as previously indicated, there is no reasonable expectation of success that block copolymers will themselves emulsify in aqueous suspension. *See, e.g.,* specification at page 47, lines 24-28.

Based on the above, as well as reasons already of record, claims 1-10, 16-28, and 30 are nonobvious under Lyons in view of either Desai or Madden, in further combination with Unger. Applicants therefore, respectfully request that this rejection be withdrawn.

CONCLUSION

In view of the above, each of the presently pending claims in this application is believed to be in immediate condition for allowance. Accordingly, the Examiner is respectfully requested to withdraw the outstanding rejection of the claims and to pass this application to issue. If it is determined that a telephone conference would expedite the prosecution of this application, the Examiner is invited to telephone the undersigned at the number given below.

In the event the U.S. Patent and Trademark office determines that an extension and/or other relief is required, applicant petitions for any required relief including extensions of time and authorizes the Commissioner to charge the cost of such petitions and/or other fees due in connection with the filing of this document to Deposit Account No. **03-1952** referencing docket no. 273012011700. However, the Commissioner is not authorized to charge the cost of the issue fee to the Deposit Account.

Dated: February 20, 2007

Respectfully submitted,

By 
Leslie A. Robinson

Registration No.: 54,403
MORRISON & FOERSTER LLP
12531 High Bluff Drive, Suite 100
San Diego, California 92130-2040
(858) 314-7692

Review

Liposomal drug delivery, a novel approach: PLARosomes^{★*}

Arkadiusz Kozubek[✉], Jerzy Gubernator, Ewa Przeworska and Maria Stasiuk

*Department of Lipids and Liposomes, Institute of Biochemistry and Molecular Biology,
University of Wrocław, St. Przybyszewskiego 63/77, 51-148 Wrocław, Poland*

Received: 31 July, 2000; accepted: 10 August, 2000

Key words: amphiphiles, vesicles, liposomes, drugs

Almost from the time of their rediscovery in the 60's and the demonstration of their entrapment potential, liposomal vesicles have drawn attention of researchers as potential carriers of various bioactive molecules that could be used for therapeutic applications in humans and animals. Several commercial liposome-based drugs have already been discovered, registered and introduced with great success on the pharmaceutical market. However, further studies, focusing on the elaboration of more efficient and stable amphiphile-based vesicular (or non-viral) drug carriers are still under investigation. In this review we present the achievements of our group in this field. We have discovered that natural amphiphilic dihydroxyphenols and their semisynthetic derivatives are promising additives to liposomal lipid compositions. The presence of these compounds in lipid composition enhances liposomal drug encapsulation, reduces the amount of the lipid carrier necessary for efficient entrapment of anthracycline drugs by a factor of two, stabilizes liposomal formulation of the drug (both in suspension and in a lyophilized powder), does not influence liposomal fate in the blood circulation system and benefits from other biological activities of their resorcinolic lipid modifiers.

Although the formation of vesicular structures from lecithin upon its hydration was demonstrated by Lehmann at the beginning of the passing century (for a review see [1]),

their potential applications were fully recognized only in the mid 60's by Bangham and his coworkers [2]. This group of researchers demonstrated that those particles, produced upon

[★]75th Anniversary of Membrane Lipid Bilayer Concept.

^{*}The support from the University of Wrocław research grants is acknowledged.

[✉]Tel.: (48 71) 324 7204; fax: (48 71) 325 2930; e-mail: kozubek@angband.microb.uni.wroc.pl

Abbreviations: AR, alkylresorcinol; LUV, large unilamellar vesicles; MLV, multilamellar vesicles; MSAR, *O*-myristoyl-*O'*-sulphoalkylresorcinol; PLARosomes, phospholipid-alkylresorcinol liposomes; SUV, small unilamellar vesicles.

swelling of a lipid film in water during agitation, sequestered part of the solution (and the solutes in it) into their interior and, what is more important, that they are surrounded by the lipid layer that functions as a permeability barrier [2]. This barrier made their properties similar to osmometers and isolated cells. These structures were finally named liposomes, in Greek "fat bodies". Since this discovery, two main streams of studies have been created, basic studies on liposomes as model membranes and research on the practical application of these structures in various aspects of human life. This second stream of studies resulted in the development of numerous small high-tech liposome-oriented pharmaceutical companies. As a result, there are already several commercially available pharmaceutical products based on drug-in-liposome formulations. Most of them concern anticancer drugs that, administered in their free form, are toxic or exhibit serious side-effects and their encapsulation into liposomal vesicles significantly diminishes these unwanted properties. In such cases, liposomes serve as a reservoir for the drug. The milestones in liposome technology were the development and introduction on the market in 1995–1997 of the liposome-based drugs: DAUNOXOME®, DOXIL® and AmBisome®. The accumulation of many novel experiences in the practical aspects of liposomes, together with new developments in basic research, will bring the field of liposome biotechnology to the place it deserves in the future.

LIPOSOMES AS DRUG DELIVERY SYSTEMS

Liposomal vesicles were prepared in the early years of their history from various lipid classes identical to those present in most biological membranes. Basic studies on liposomal vesicles resulted in numerous methods of their preparation and characterization. Liposomes are broadly defined as lipid

bilayers surrounding an aqueous space. Multilamellar vesicles (MLV) consist of several (up to 14) lipid layers (in an onion-like arrangement) separated from one another by a layer of aqueous solution. These vesicles are over several hundred nanometers in diameter. Small unilamellar vesicles (SUV) are surrounded by a single lipid layer and are 25–50 nm (according to some authors up to 100 nm) in diameter. Large unilamellar vesicles (LUV) are, in fact, a very heterogeneous group of vesicles that, like the SUVs, are surrounded by a single lipid layer. The diameter of these liposomes is very broad, from 100 nm up to cell size (giant vesicles) [3]. Besides the technique used for their formation the lipid composition of liposomes is also, in most cases, very important. For some bioactive compounds the presence of net charged lipids not only prevents spontaneous aggregation of liposomes but also determines the effectiveness of the entrapment of the solute into the liposomal vesicles. Natural lipids, particularly those, with aliphatic chains attached to the backbone by means of ester or amide bonds (phospholipids, sphingolipids and glycolipids) are often subject to the action of various hydrolytic (lipolytic) enzymes when injected into the animal or human body. These enzymes cleave off acyl chains and the resulting lysolipids have destabilising properties for the lipid layer and cause the release of the entrapped bioactive component(s). As a result new types of vesicles, that should merely bear the name of liposomes as their components are lipids only by similarity of their properties to natural (phospho)lipids, have been elaborated. These vesicles, still named liposomes, are made of various amphiphile molecules (the list of components is long). The crucial feature of these molecules is that upon hydration they are able to form aggregation structures resembling an array and have properties of natural phospholipid bilayers. Among such molecules, various amphiphiles have been employed, such as ether lipids [4–7], fluorinated lipid [8–10], synthetic dou-

ble chain amphiphiles [11] as well as single chain amphiphiles, such as N-alkylindoles [12], polyhydroxyl lipids [13], polyhedral non-ionic surfactants [11, 14-16], polymerized liposomes [17, 18], cationic amphiphiles [16], plasmalogens [19] and others [20-27]. This resulted in various new formulations of vesicle compositions and new names given to them, such as niosomes, letherosomes, archeosomes, etc.

The common feature of classical liposomes, i.e., made preferentially of phospholipids, and of vesicles made of amphiphilic molecules, was their ability to form dynamic lamellar structures with barrier properties separating the interior of the vesicles from the outside medium. One may conclude that, at present, the term "liposomes" covers not only phospholipid-based vesicles but also other vesicular structures with properties identical or similar to those of classical, natural phospholipid based liposomes.

In the early 70's the use of liposomes as a drug carrier system was proposed by Gregoriadis & Ryman [28]. Since this first report, liposomes were developed as an advanced drug delivery vehicle. They are generally considered non-toxic, biodegradable and non-immunogenic. Associating a drug with liposomes markedly changes its pharmacokinetics and lowers systemic toxicity; furthermore, the drug is prevented from early degradation and/or inactivation after introduction to the target organism [29-34].

The use of liposomes or, in general, vesicular structures for the delivery of various active compounds is recognized in relation to water solubility of the compound. When the compound is water soluble, the size and volume of the aqueous compartment of the vesicle is crucial. In contrast, hydrophobic compounds will prefer incorporation into the lipid (amphiphile) layer that constructs the vesicle. In such a case, the size of the aqueous compartment is not important. Depending on the need, one can use SUV type or MLV type vesicles for effective entrapment and delivery of

the drug to the target tissues or cells [35, 36]. Nevertheless, charge properties and interactions of the active compound with vesicle forming molecules will determine the effectivity of entrapment, i.e., the amount of the compound that can be "loaded" into a single vesicle [37-40]. On the other hand, the composition of the molecules used for the formation of the vesicular structure will, at least, affect the fate of vesicles from the site of their introduction as well as the interaction with components of the body (e.g., surface charge [37-39], serum proteins, lipoproteins, opsonin system [40, 41], phagocytic system [42] and finally target cells [41, 43, 44]). In the earlier studies, when therapeutically active substance were not easily available, most of the experiments were done using a marker compound. The results, however, were not the same as those obtained in experiments in which an active substance was used and the conditions were more related to the real situation (*ex vivo*, *in vivo*). These findings implicate the necessity for studies in which an active substance is used and the conditions of the experiments resemble, as closely as possible, those of therapeutic liposomal (vesicular) drug application.

The benefits of liposomal formulations were already demonstrated clinically and stimulate many laboratories (research and pharmaceutical) in their efforts to introduce new liposomal/vesicular drugs. These can be illustrated with the data presented in Table 1.

A NOVEL APPROACH TO LIPOSOMAL ANTHRACYCLINE DRUGS: PLARosomes

In contrast to the thoroughly studied application of various derivatives of lipid and synthetic amphiphiles for the encapsulation of various drugs [9, 12, 13, 15, 23, 45-50], the idea of modifying the vesicle's lipid barrier/encapsulation properties by means of lipid layer modifying molecules is relatively

Table 1. Examples of drugs in liposomal formulations

Drug	Application	Commercial name	Composition of liposomes
Amikacin	Bacterial infections	MiKasome	HSPC/CH/DSPG
Adriamycin (doxorubicin)	Stomach cancer	-	DPPC/CH
Ampicilin	Listeria monocytogenes	-	CH/PC/PS 5:4:1 CH:DSPC:DPPG 10:10:1
Annamycin	Kaposi's sarcoma, Breast cancer, Leukemia	Annamycin	Liposomes
Amphotericin B	Systemic fungal infections	AmBisome	HSPC/CH/DSPG
All- <i>trans</i> -retinoic acid	Acute promyelocytic leukemia, Lymphoma, Prostate cancer	ATRAGEN	Liposomes
Muramyl dipeptide	Immunostimulator	-	DSPC/PS 1:1
1- β -D-Arabinofuranozidecytosine	Leukemia	-	SM/PC/CH 1:1:1
Ciprofloxacin	<i>Pseudomonas aeruginosa</i>	-	DPPC
Clodronate	Macrophage suppression	-	PC/CH
Cis-diaminodichloroplatinum(II)	Cancers	-	Liposomes
Cyclosporin	Immunosuppressor	-	PC/CH
Chloroquine	Malaria	-	PC/PG/CH 10:1:5
Cu/Zn superoxide dismutase	Antiinflammatory	-	Liposomes
Doxorubicin	Cancers	Doxil	HSPC/CH/PEG-DSPE
Doxorubicin	Breast cancer	EVACET	
Daunorubicin	Cancers	DaunoXome	DSPC/CH
Ganciclovir	Cytomegalovirus retinitis	-	Liposomes
Interleukin 2	Immunostimulant	-	DMPC
Leukotriene A4	Not estimated	-	PC/DCP/CH 7:2:1
Lipid A	Immunoadjuvant	-	Liposomes
Mitoxantron	Colon cancer	-	PC/Ch 7:1
Methotrexate	Cancers	-	DPPC/PI 18:2w/w
Nystatin	Systemic fungal infections	NYOTRAN	Liposomes
Na ₃ (B ₂₀ H ₁₇ NH ₃)	Cancers	-	DSPC/CH
Pentostam	Leishmanioses	-	Niosomes
Platinum drugs e.g., cisplatin	Mezotelioma	PLATAR	Liposomes
Lurtotecan	Cancers	NX 211	Liposomes
Oligonucleotides against <i>c-myc</i>	Cancers	INXC-6295	Liposomes TCS
Prostaglandin E1	Anti-inflammatory	-	PC
Ribavirin	Herpes simplex	-	Liposomes
Streptosotocin	Lymphocyte activator	-	DMPC/CH 2:1
Suramin	Trypanosomes	-	DPPC

Muramyl tripeptide	Immunostimulant		Liposomes
Ether lipids	Non-small-cell lung cancer, Melanoma, Leukemia, Human prostate cancer	TLC ELL12	Liposomes
Vincristin	Cancers	VincaXome	DSPC/CH
Vincristin	Cancers, Lymphoma	Onco TCS	Liposomes TCS
Various drugs and contrasts	Diagnostics of various diseases	lipoMASC	PEG-liposomes

Abbreviations to Table 1: HSPC, hydrogenated soya phosphatidylcholine (hydrogenated soya lecithin); CH, cholesterol; DSPG, distearoylphosphatidylglycerol; DPPG, dipalmitoylphosphatidylglycerol; DMPC, dimyristoylphosphatidylcholine; DPPC, dipalmitoylphosphatidylcholine; DSPC, distearoylphosphatidylcholine; PC, phosphatidylcholine; PS, phosphatidylserine; SM, sphingomyelin; PI, phosphatidylinositol; DCP, dicetylphosphate; PEG-DSPE, polyethylene glycol-phosphatidylethanolamine derivative; liposomes TCS, commercial composition of liposome-forming lipids; PEG-liposomes, liposomes modified with components containing a PEG (polyethylene glycol); Liposomes, liposomes with composition not defined.

less exploited. Our research focusses on the biological properties of natural single chain amphiphilic compounds belonging to the group of non-isoprenoic phenolic lipids [51]. They are amphiphilic in nature due to the non-isoprenoid side chains attached to the hydroxybenzene ring and are believed to be derived from the polyketide (acetate) pathway, as for example 6-pentadecylsalicylic acid. Resorcinolic lipids, alternatively called alkyl-resorcinols or 5-alkylresorcinols that are derivatives of resorcinol or higher homologs of orcinol (1,3-dihydroxy-5-methylbenzene), are of interest from the biopharmacological, biomedical and biotechnological points of view. The occurrence in their molecules of a benzene ring with two hydrophilic OH attachments and long hydrocarbon chains (in most cases C15–C21) determines their ability to interact strongly with biological membranes and phospholipid bilayers.

We have demonstrated that these compounds act differentially upon lipid bilayer permeability depending on their localization [52]. When they were introduced into a suspension of liposomes made of phospholipids, they caused an increase of membrane permeability and the release of liposomal contents. On the other hand, when introduced into the lipid mixture prior to the formation of the liposomal structures, they induced stabilization of the liposomal membrane, similarly to the effect observed for cholesterol [52, 53].

Furthermore, an enhancement of liposomal entrapment and simultaneously a decrease of vesicle diameter were observed [52, 54, 55]. The entrapment of various markers (Patent Blue Violet, 5,6-carboxyfluorescein, calcein, potassium chromate) into liposomes made of phosphatidylcholine, phosphatidylethanolamine, their mixtures and mixtures of phosphatidylcholine/cholesterol varied significantly. It was dependent both on the amount of the modifier in the liposomal lipid mixture (at 10% of C15 resorcinolic lipid a twofold increase of captured volume was already noted) and on the length of the aliphatic chain. The most effective were 5-*n*-nonadecylresorcinol and resorcinolic lipids isolated from cereal bran with mean chain length of about C18. It should be pointed out that the similarity of the resorcinolic lipid composition between natural preparations was very high, as it has been determined. Liposomal vesicles obtained from phospholipid-resorcinolic lipids mixtures only by application of the freezing and thawing were stable and of mean diameter 250–280 nm. This stability is probably due to the inverted cone-like shape of a resorcinolic lipid molecule and the related increase of the bilayer curvature. Liposomes modified with resorcinolic lipids, are easily calibrated by extrusion through polycarbonate membranes and their size remains stable (within 25%) over weeks, both at 4°C and at room temperature. Modified liposomal vesicles composed of

lecithin, phosphatidylethanolamine or mixtures of these lipids were also stable with respect to the leakage of their contents and they did not differ significantly from phosphatidylcholine/cholesterol (50:50, molar ratio) liposomes. Generally, in terms of stability of size and contents, the resorcinolic lipids-modified liposomes behaved similarly to conventional phospholipid or phospholipid/cholesterol vesicles *in vitro* and *in vivo* [54]. A similar tendency of resorcinolic lipids was observed when sphingolipids (ceramides, sphingomyelin), cholesterol and free fatty acids (palmitic and linoleic) containing liposomes were studied [56].

In another approach, we have demonstrated that resorcinolic lipids at high pH form, alone or in mixtures with free fatty acids or cholesterol, vesicular structures (110–190 nm mean diameter) of captured volume varying from 4–8 L/mol. The size of most of the vesicles was stable at 4°C for at least 3–6 months [56].

According to this, and the results mentioned earlier, the introduction of negative charges into the liposomal membrane seems to be one of the crucial factors governing the effective encapsulation and binding of anthracycline drugs (one of the most exploited and used anticancer drugs). A derivative of resorcinolic lipid with a strong negative charge was synthesised. To change the inverted conical shape of the molecule, one of the hydroxyl groups was esterified with myristoyl chloride. The second hydroxyl group was modified by sulfonation, thus a strong negative charge was introduced. This compound we named MSAR (*O*-Myristoyl-*O'*-Sulfo-AlkylResorcinol) [54, 55]. This compound, as expected, formed, upon hydration, vesicular structures capable of efficient entrapment of the solutes present in the aqueous phase. The introduction of this compound into the liposome-forming lipid mixture containing egg phosphatidylcholine and hydrogenated phosphatidylcholine, at a level not exceeding 50% (w/w), resulted in a several-fold enhancement of the liposomal drug's encapsulation. The encapsulation of

both markers as well as the anthracycline drugs (doxorubicin, mitoxantrone) was higher than in the case of conventional liposomes – the drug to lipid ratio was lowered from 15 to 8 with the encapsulation efficiency between 90 and 99% [54]. For such a high encapsulation efficiency no additional procedures such as gradient loading, is needed. These new liposomes showed enhanced shelf-life and stability of size both at room temperature and at about 4°C. Additionally, formulation of anthracycline drugs in modified liposomes allowed the production of the dry (lyophilized) form of the active substance with size and stability after rehydration identical to those before this process (Gubernator & Kozubek, unpublished). Therefore, we have named vesicles modified with resorcinolic lipids or their derivatives PLARosomes abbreviated from PhosphoLipid-AlkylResorcinol liposomes.

PLARosomes, when incubated with human plasma, were more stable than conventional lecithin/cholesterol or sphingomyelin/cholesterol liposomes. Conventional liposomes, in a two-week experiment, released up to 50% of their contents whereas PLARosomes, despite their composition, released only 3–20% of their contents. Liposomes modified with alkylresorcinol (AR) or MSAR are cleared from the circulation in a similar way as conventional liposomes. However, the clearance of PLARosomes with high load of MSAR (30%, w/w) from the circulation was drastically enhanced by their strong negative charge [56]. This results in their preferential localization in the liver (up to 60%, 24 h after injection). Similar liposomes, but modified with ARs, were found in almost equal amounts in the liver and spleen. Vesicles containing less modifier did not differ in their distribution from sphingomyelin/cholesterol liposomes, belonging to the longest circulating conventional liposomes. Approximately 12% of both types of liposomes were still present in the blood 24 h after injection [56].

What other benefits could be related to the new liposomes? Conventional liposomes, be-

sides their uptake by the reticulo-endothelial macrophage system, are also subject to the action of various lipolytic plasma enzymes, including phospholipase A₂. This enzyme alone will alter the barrier properties of the lipid bilayer by hydrolysis of the phospholipid components into lysophospholipids, responsible for the destruction of bilayer integrity. Resorcinolic lipids present in the bilayer alter phospholipase kinetics so that they may be considered functional inhibitors of the enzyme [57]. These properties might be responsible for the protection of liposomes from the action of these enzymes. Additionally, resorcinolic lipids, due to their phenolic nature, exhibit antioxidant properties [54, 58–60]. The use of natural antioxidants for the protection and stabilization of liposomal components against oxidative damage has been reported recently [49, 61]. Resorcinolic lipids, in the context of the above findings, would also play a role in such protection. On the other hand, resorcinolic lipids, besides their direct or indirect effect upon liposomal drug-entrapping properties, display other properties crucial to the human organism. These compounds have been demonstrated as having antimutagenic properties [62], especially when metabolically activated mutagenic compounds are considered (e.g., benzopyrene) [63, 64].

Additionally, resorcinolic lipids are effective functional inhibitors of triglyceride synthesis and the enzymatic oxidation of unsaturated fatty acids [65, 66], which makes possible a synergistic effect of those compounds on the organisms into which they were introduced. Recent reports indicating the participation of resorcinolic lipids in DNA damage [67–71] and inhibition of its repair [72, 73] with parallel enhancement of anticancer drugs activity, strongly support the necessity for further studies of this interesting group of natural compounds. The concomitant biological activities may be a key issue in creating a kind of universal vesicular systems based on natural and biodegradable modifiers. It may also be

speculated that resorcinolic lipids, the compounds present in all whole grain-based products, and used for centuries in human nutrition, will participate in establishing a modern way of liposome medical applications.

CONCLUDING REMARKS

The ability of liposomes consisting of components other than phospholipids and cholesterol or their semisynthetic derivatives to enhance the encapsulation of bioactive substances provides new promising perspectives for establishing new, efficient and stable carriers for drug delivery. We hope that in the near future PLARosomes, the components of which are not directly toxic (unpublished data), will be used for the efficient entrapment and delivery of drugs to human or animal organisms. Resorcinolic lipids and modern studies on their biological activities are relatively new but show a tremendous potential not only as components of PLARosomes [72, 73]. PLARosomes may come into commercial use relatively soon because of the benefits anticipated from the knowledge generated through the use of conventional liposomes.

We acknowledge the contributions of the staff of our resorcinolic lipids project team at the Institute of Biochemistry and Molecular Biology of the University of Wrocław.

REFERENCES

1. Lasic, D.D. (1993) *Liposomes: from Physics to Applications*. Elsevier, Amsterdam, London, New York.
2. Bangham, A.D., Standish, M.M. & Watkins, J.C. (1965) Diffusion of univalent ions across lamellae of swollen phospholipids. *J. Mol. Biol.* 13, 238–252.

3. Woodle, M.C. & Papahadjopoulos, D. (1989) Liposome preparation and size characterization. *Methods Enzymol.* **171**, 193–217.
4. Choquet, C.G., Patel, G.B., Beveridge, T.J. & Sprott, G.D. (1994) Stability of pressure extruded liposomes made from archaeobacterial ether lipids. *Appl. Microbiol. Biotechnol.* **42**, 375–384.
5. Patel, G.B. & Sprott, G.D. (1999) Archaeobacterial ether lipid liposomes (Archaeosomes) as novel vaccine and drug delivery systems. *Crit. Rev. Biotechnol.* **19**, 317–357.
6. Sprott, G.D., Tolson, D.L. & Patel, G.B. (1997) Archaeosomes as novel antigen delivery systems. *FEMS Microbiol. Lett.* **154**, 17–22.
7. Yamauchi, K., Doi, K. & Kinoshita, M. (1996) Archaeobacterial lipid models: Stable liposomes from 1-alkyl-2-phytan-yl-sn-glycero-3-phosphocholines. *Biochim. Biophys. Acta* **1283**, 163–169.
8. Clary, L., Verderone, G., Santaella, C. & Vierling, P. (1997) Membrane permeability and stability of liposomes made from highly fluorinated double-chain phosphocholines derived from diaminopropanol, serine or ethanolamine. *Biochim. Biophys. Acta* **1328**, 55–64.
9. Gadras, C., Santaella, C. & Vierling, P. (1999) Improved stability of highly fluorinated phospholipid-based vesicles in the presence of bile salts. *J. Control. Rel.* **57**, 29–34.
10. Riess, J.G. (1994) Fluorinated vesicles. *J. Drug Target.* **2**, 455–468.
11. Pector, V., Caspers, J., Banerjee, S., Derie-maeker, L., Fuks, R., ElOuahabi, A., Vandenbranden, M., Finsy, R. & Ruysschaert, J.M. (1998) Physico-chemical characterization of a double long-chain cationic amphiphile (Vectamidine) by microelectrophoresis. *Biochim. Biophys. Acta* **1372**, 339–346.
12. Abel, E., Fedders, M.F. & Gokel, G.W. (1995) Vesicle formation from N-alkylindoles: Implications for tryptophan water interactions. *J. Am. Chem. Soc.* **117**, 1265–1270.
13. Assadullahi, T.P., Hider, R.C. & McAuley, A.J. (1991) Liposome formation from synthetic polyhydroxyl lipids. *Biochim. Biophys. Acta* **1083**, 271–276.
14. Arunothayanun, P., Uchegbu, I.F. & Florence, A.T. (1999) Osmotic behaviour of polyhedral non-ionic surfactant vesicles (niosomes). *J. Pharm. Pharmacol.* **51**, 651–657.
15. Baillie, A.J., Coombs, G.H., Dolan, T.F. & Laurie, J. (1986) Non-ionic surfactant vesicles, niosomes, as a delivery system for the anti-leishmanial drug, sodium stibogluconate. *J. Pharm. Pharmacol.* **38**, 502–505.
16. Budker, V., Gurevich, V., Hagstrom, J.E., Bortzov, F. & Wolff, J.A. (1996) pH-Sensitive, cationic liposomes: A new synthetic virus-like vector. *Nature Biotechnol.* **14**, 760–764.
17. Chen, H.M., Torchilin, V. & Langer, R. (1996) Lectin-bearing polymerized liposomes as potential oral vaccine carriers. *Pharm. Res.* **13**, 1378–1383.
18. Chen, H.M., Torchilin, V. & Langer, R. (1996) Polymerized liposomes as potential oral vaccine carriers: Stability and bioavailability. *J. Control. Rel.* **42**, 263–272.
19. Thompson, D.H., Gerasimov, O.V., Wheeler, J.J., Rui, Y.J. & Anderson, V.C. (1996) Triggerable plasmalogen liposomes: Improvement of system efficiency. *Biochim. Biophys. Acta* **1279**, 25–34.
20. Cistola, D.P., Atkinson, D., Hamilton, J.A. & Small, D.M. (1986) Phase behavior and bilayer properties of fatty acids:hydrated 1:1 acid-soaps. *Biochemistry* **25**, 2804–2812.
21. Hu, C.J. & Rhodes, D.G. (1999) Proniosomes: A novel drug carrier preparation. *Int. J. Pharm.* **185**, 23–35.
22. Murdan, S., Gregoriadis, G. & Florence, A.T. (1999) Sorbitan monostearate polysorbate 20 organogels containing niosomes: A delivery

- vehicle for antigens? *Eur. J. Pharm. Sci.* **8**, 177-185.
23. Uchegbu, I.F. & Duncan, R. (1997) Niosomes containing *N*-(2-hydroxypropyl)methacrylamide copolymer-doxorubicin (PK1): Effect of method of preparation and choice of surfactant on niosome characteristics and a preliminary study of body distribution. *Int. J. Pharm.* **155**, 7-17.
24. Uchegbu, I. (1998) The biodistribution of novel 200-nm palmitoyl muramic acid vesicles. *Int. J. Pharm.* **162**, 19-27.
25. Uchegbu, I.F., Schatzlein, A.G., Tetley, L., Gray, A.I., Sludden, J., Siddigne, S. & Mosha, E. (1998) Polymeric chitosan-based vesicles for drug delivery. *J. Pharm. Pharmacol.* **50**, 453-458.
26. Vora, B., Khopade, A.J. & Jain, N.K. (1998) Proniosome based transdermal delivery of levonorgestrel for effective contraception. *J. Control. Rel.* **54**, 149-165.
27. Weissig, V., Lasch, J., Erdos, G., Meyer, H.W., Rowe, T.C. & Hughes, J. (1998) DQAsomes: A novel potential drug and gene delivery system made from Dequalinium(TM). *Pharm. Res.* **15**, 334-337.
28. Gregoriadis, G. & Ryman, B.E. (1972) Lysosomal localization of fructofuranoside-containing liposomes injected into rats. *Biochem. J.* **129**, 123-133.
29. Allen, T.M. (1997) Liposomes: Opportunities in drug delivery. *Drugs* **54**, 8-14.
30. Allen, T.M. & Moase, E.H. (1996) Therapeutic opportunities for targeted liposomal drug delivery. *Adv. Drug Deliv. Rev.* **21**, 117-133.
31. Bally, M.B., Nayar, R., Masin, D., Hope, M.J., Cullis, P.R. & Mayer, L.D. (1990) Liposomes with entrapped doxorubicin exhibit extended blood residence times. *Biochim. Biophys. Acta* **1023**, 133-139.
32. Bandak, S., Ramu, A., Barenholz, Y. & Gabizon, A. (1999) Reduced UV-induced degradation of doxorubicin encapsulated in polyethyleneglycol-coated liposomes. *Pharm. Res.* **16**, 841-846.
33. Coukell, A.J. & Spencer, C.M. (1997) Polyethylene glycol-liposomal doxorubicin: A review of its pharmacodynamic and pharmacokinetic properties, and therapeutic efficacy in the management of AIDS-related Kaposi's sarcoma. *Drugs* **53**, 520-538.
34. Gabizon, A., Goren, D., Cohen, R. & Barenholz, Y. (1998) Development of liposomal anthracyclines: From basics to clinical applications. *J. Control. Rel.* **53**, 275-279.
35. Kulkarni, S.B., Betageri, G.V. & Singh, M. (1995) Factors affecting microencapsulation of drugs in liposomes. *J. Microencapsul.* **12**, 229-246.
36. Nagayasu, A., Uchiyama, K. & Kiwada, H. (1999) The size of liposomes: A factor which affects their targeting efficiency to tumors and therapeutic activity of liposomal antitumor drugs. *Adv. Drug Deliv. Rev.* **40**, 75-87.
37. Bajoria, R. & Contractor, S.F. (1997) Effect of surface charge of small unilamellar liposomes on uptake and transfer of carboxyfluorescein across the perfused human term placenta. *Pediatr. Res.* **42**, 520-527.
38. Miller, C.R., Bondurant, B., McLean, S.D., McGovern, K.A. & O'Brien, D.F. (1998) Liposome-cell interactions in vitro: Effect of liposome surface charge on the binding and endocytosis of conventional and sterically stabilized liposomes. *Biochemistry* **37**, 12875-12883.
39. Nakanishi, T., Kunisawa, J., Hayashi, A., Tsutsumi, Y., Kubo, K., Nakagawa, S., Fujiwara, H., Hamaoka, T. & Mayumi, T. (1997) Positively charged liposome functions as an efficient immunoadjuvant in inducing immune responses to soluble proteins. *Biochem. Biophys. Res. Commun.* **240**, 793-797.
40. Scherphof, G.L. & Kamps, J.A.A.M. (1998) Receptor versus non-receptor mediated clear-

- ance of liposomes. *Adv. Drug Deliv. Rev.* **32**, 81-97.
41. Papisov, M.I. (1998) Theoretical considerations of RES-avoiding liposomes: Molecular mechanics and chemistry of liposome interactions. *Adv. Drug Deliv. Rev.* **32**, 119-138.
42. Bakker, J., Sanders, A. & Van Rooijen, N. (1998) Effects of liposome-encapsulated drugs on macrophages: Comparative activity of the diamidine 4',6-diamidino-2-phenylindole and the phenanthridinium salts ethidium bromide and propidium iodide. *Biochim. Biophys. Acta* **1373**, 93-100.
43. Mayer, L.D. (1998) Future developments in the selectivity of anticancer agents: Drug delivery and molecular target strategies. *Cancer Metastasis Rev.* **17**, 211-218.
44. Sharma, A. & Sharma, U.S. (1997) Liposomes in drug delivery: Progress and limitations. *Int. J. Pharm.* **154**, 123-140.
45. Engberts, J.B.F.N. & Hoekstra, D. (1995) Vesicle-forming synthetic amphiphiles. *Biochim. Biophys. Acta* **1241**, 323-340.
46. Gluck, R. & Wegmann, A. (1997) Virosomes, a new liposome-like vaccine delivery system; in *Antigen Delivery Systems* (Gander, B., Merkle, H.P. & Corradin, G., eds.) pp. 101-122, Harwood Academic Publ.
47. Han, S.K., Ko, Y.I., Park, S.J., Jin, I.J. & Kim, Y.M. (1997) Oleanolic acid and ursolic acid stabilize liposomal membranes. *Lipids* **32**, 769-773.
48. Shimizu, K., Maitani, Y., Takayama, K. & Nagai, T. (1997) Formulation of liposomes with a soybean-derived sterylglucoside mixture and cholesterol for liver targeting. *Biol. Pharm. Bull.* **20**, 881-886.
49. Waters, R.E., White, L.L. & May, J.M. (1997) Liposomes containing alpha-tocopherol and ascorbate are protected from an external oxidant stress. *Free Radic. Res.* **26**, 373-379.
50. Zeisig, R., Arndt, D., Stahn, R. & Fichtner, I. (1998) Physical properties and pharmacological activity *in vitro* and *in vivo* of optimised liposomes prepared from a new cancerostatic alkylphospholipid. *Biochim. Biophys. Acta* **1414**, 238-248.
51. Kozubek, A. & Tyman, J.H.P. (1999) Resorcinolic lipids, the natural non-isoprenoid phenolic amphiphiles and their biological activity. *Chem. Rev.* **99**, 1-26.
52. Gubernator, J., Stasiuk, M. & Kozubek, A. (1999) Dual effect of alkylresorcinols, natural amphiphilic compounds, upon liposomal permeability. *Biochim. Biophys. Acta* **1418**, 253-260.
53. Kozubek, A., Nietubyc, M. & Sikorski, A.F. (1992) Modulation of the activities of membrane enzymes by cereal grain resorcinolic lipids. *Z. Naturforsch.* **47c**, 41-46.
54. Gubernator, J. (1998) Modification of the captured volume and the stability of liposomes as drug carriers by resorcinolic lipids and their derivatives. *PhD Thesis*, Department of Lipids and Liposomes, University of Wrocław, Wrocław.
55. Gubernator, J. & Kozubek, A. (1998) Liposomes modified with alkylresorcinols. *Chem. Phys. Lipids* **94**, 177.
56. Przeworska, E. (2000) The properties of liposomes containing resorcinolic lipids. *PhD Thesis*, Department of Lipids and Liposomes, University of Wrocław, Wrocław.
57. Kozubek, A. (1992) The effect of resorcinolic lipids on phospholipid hydrolysis by phospholipase A2. *Z. Naturforsch.* **47c**, 608-612.
58. Kozubek, A. & Nienartowicz, B. (1995) Cereal grain resorcinolic lipids inhibit H₂O₂-induced peroxidation of biological membranes. *Acta Biochim. Polon.* **42**, 309-316.
59. Nienartowicz, B. & Kozubek, A. (1993) Antioxidant activity of cereal bran resorcinolic lipids. *Pol. J. Food Nutr. Sci.* **2**, 51-60.

60. Struski, D.G.J. & Kozubek, A. (1992) Cereal grain alk(en)ylresorcinols protect lipids against ferrous ions-induced peroxidation. *Z. Naturforsch.* **47c**, 47-50.
61. Woodall, A.A., Britton, G. & Jackson, M.J. (1997) Carotenoids and protection of phospholipids in solution or in liposomes against oxidation by peroxy radicals: Relationship between carotenoid structure and protective ability. *Biochim. Biophys. Acta* **1336**, 575-586.
62. George, J. & Kuttan, R. (1997) Mutagenic, carcinogenic and cocarcinogenic activity of cashewnut shell liquid. *Cancer Lett.* **112**, 11-16.
63. Gasiorowski, K., Szyba, K., Brokos, B. & Kozubek, A. (1996) Antimutagenic activity of alkylresorcinols from cereal grains. *Cancer Lett.* **106**, 109-115.
64. Gasiorowski, K., Brokos, B., Kozubek, A. & Oszmianski, J. (2000) The antimutagenic activity of two plant-derived compounds. A comparative cytogenetic studies. *Cell Mol. Biol. Lett.* **5**, 171-190.
65. Rejman, J. & Kozubek, A. (1997) Long-chain orcinol homologs from cereal bran are effective inhibitors of glycerophosphate dehydrogenase. *Cell Mol. Biol. Lett.* **2**, 411-419.
66. Deszcz, L. & Kozubek, A. (1997) Inhibition of soybean lipoxygenases by resorcinolic lipids from cereal bran. *Cell. Mol. Biol. Lett.* **2**, 213-222.
67. Barr, J.R., Murty, V.S., Yamaguchi, K., Singh, S., Smith, D.H. & Hecht, S.M. (1988) 5-Alkylresorcinols from *Hakea amplexicaulis* that cleave DNA. *Chem. Res. Toxicol.* **1**, 204-207.
68. Hecht, S.M. (1989) Natural products that cleave DNA. *Pure Appl. Chem.* **61**, 577-580.
69. Lytollis, W., Scannell, R.T., An, H.Y., Murty, V.S., Reddy, K.S., Barr, J.R. & Hecht, S.M. (1995) 5-Alkylresorcinols from *Hakea trifurcata* that cleave DNA. *J. Am. Chem. Soc.* **117**, 12683-12690.
70. Nagai, K., Carter, B.J., Xu, J. & Hecht, S.M. (1991) DNA cleavage by oxygen radicals produced in the absence of metal ions or light. *J. Am. Chem. Soc.* **113**, 5099-5100.
71. Scannell, R.T., Barr, J.R., Murty, V.S., Reddy, K.S. & Hecht, S.M. (1988) DNA strand scission by naturally occurring 5-alkylresorcinols. *J. Am. Chem. Soc.* **110**, 3650-3651.
72. Deng, J.Z., Starck, S.R. & Hecht, S.M. (1999) Bis-5-alkylresorcinols from *Panopsis rubescens* that inhibit DNA polymerase beta. *J. Nat. Prod.* **62**, 477-481.
73. Starck, S.R., Deng, J.Z. & Hecht, S.M. (2000) Naturally occurring alkylresorcinols that mediate DNA damage and inhibit its repair. *Biochemistry* **39**, 2413.